#### **Unit Conversions**

\*You can multiply anything by 1 without changing the physical meaning\*

## **How Do We Multiply By 1**

If the same thing is on the top and the bottom of a fraction,

Then they cancel out... and become 1.

## All Equalities = 1

Any time you put two things that are equal to each other 1000g = 1kg

On top and on bottom

Then they cancel out... and become 1.

It doesn't matter which is on top and which is on bottom.

Both cancel out... and become 1.

#### The Rules: Picket Fence Method

- Write down "given" + unit
- Draw a horizontal & vertical line "picket fence"
- Write "wanted" / "given" on the fence
- Enter proper #s for "wanted units & "given" units
- Cancel units & do the math

#### For example: Convert 25 km to meters (m)

Use the conversion factor equals one and that allows units to cancel.

### **Examples:**

### How old are you in days?

Multiply by conversion factors that equal one and allow units to cancel

(given ) 15 <del>years</del>	365 days	= 5,475 days (units wanted)	
	<del>year</del>		

# How much does a 120 lb person weigh in kilograms. There are 0.454 kg in one pound

Multiply by conversion factors that equal one and allow units to cancel

(given ) 120 <del>lbs</del>	0.454 kg	= 54.58 kgs (units wanted)	
	<del>lb</del>		

# **How many inches are in 5 yards?** (a yard is 3 feet and there are 12 inches in a foot) Multiply by conversion factors that equal one and allow units to cancel

(given ) 5 <del>yards</del>	3 <del>feet</del>	12 <del>inches</del>	= 180 inches (units wanted)
	<del>yard</del>	foot	

## How many seconds are in 7 days?

Multiply by conversion factors that equal one and allow units to cancel

(given ) 7 <del>days</del>	24 <del>hours</del>	60 minutes	60 seconds	= 604,800 seconds (units wanted)
	<del>day</del>	hour	minute	

A chemist performs the synthesis of sodium chloride from its elements. The chemist begins with 46 grams of sodium. How many moles of chlorine  $(Cl_2)$  are needed?

2 Na + 
$$Cl_2 \rightarrow 2$$
 NaCl

This balanced equation tells us that <u>one mol of  $Cl_2$ </u> is needed to react with every <u>2 moles</u> <u>of sodium (Na)</u>.

Multiply by conversion factors that equal one and allow units to cancel

(given ) 46 <del>grams Na</del>	1 <del>mol Na</del>	1 mol Cl <sub>2</sub>	= 1 mol Cl <sub>2</sub> (units wanted)
	23 <del>grams</del> <del>Na</del>	2-mol Na	

How many grams of water can be prepared from 5 moles of hydrogen at standard conditions?

$$2 H_2(g) + O_2(g) \rightarrow 2 H_2O$$

This balanced equation tells us that <u>two moles of  $H_2O$ </u> can be prepared from <u>2 moles of hydrogen ( $H_2$ ) and 1 mole of oxygen ( $O_2$ ).</u>

Multiply by conversion factors that equal one and allow units to cancel

(given ) 5 <del>mol H</del> <sub>2</sub>	2 mol H <sub>2</sub> O	18.02 g H <sub>2</sub> O	=90.1 g H <sub>2</sub> O (units wanted)
	2 mol H <sub>2</sub>	1 <del>-mol H<sub>2</sub>O</del>	